

SUBSTANCE IDENTITY PROFILE (SIP)

NICKEL SULPHATE

EC #	232-104-9					
CAS #	7786-81-4 (primary); 10101-97-0					
Other identifiers	Nickel sulfate Nickel (2+) sulfate Nickel (2+) sulphate Sulfuric acid, nickel (2++) salt (1:1) Sulfuric acid, nickel (2+) salt (1:1)		lickelous sulphate lickelous sulfate lickel (II) sulphate lickel (II) sulfate lickel sulfate (1:1) lickel sulphate (1:1)	Nickel mo Nickel mo Nickel sul Nickel sul	Nickel monosulphate Nickel monosulfate Nickel sulphate hexahydrate Nickel sulfate hexahydrate	
	Substance compositional information:					
			Compositions			
	Constituents/Impurities	Nickel Sulphate (anhydrous form)***	Nickel Sulphate (hydrates)	Nickel Sulphate (anhydrous form with arsenic impurities)	Nickel Sulphate (hydrates with arsenic impurities)	
	Nickel sulphate** EC# 232-104-9 CAS# 7786-81-4	≥90 – ≤100%	>77 – ≤100%	>77 - <100%	≥75 - <100%	
	Copper sulphate EC# 231-847-6 CAS# 7758-98-7	-	>0.0001 - <5%	≥0 - <10%	≥0 - <10%	
	Zinc sulphate EC# 231-793-3 CAS# 7733-02-0	-	>0.0001 - <3%	≥0 - <3%	≥0 - <3%	
	Arsenic acid EC# 231-901-9 CAS# 7778-39-4	-	-	>0 - <2%	>0 - ≤3%	
	Calcium sulfate EC# 231-900-3 CAS# 7778-18-9	-	-	≥0 - <9%	≥0 - <8%	
	Ferric oxide EC# 215-168-2 CAS# 1309-37-1	-	-	≥0 - <3%	≥0 - <3%	
	Other impurities*	≥0 - ≤10%	≥0 – ≤23%	≥0-<2%	≥0 - ≤25%	

*Other impurities which are not relevant for the classification and/or for PBT assessment and that are each of them present in a concentration <1%.

**In the hydrated compositions, the nickel sulphate concentration was calculated from results of continuous analyses and refers to NiSO4 (including crystalline water). Although there is a small deviation from the 80/20 rule for the mono-constituent substance, the typical concentration is above 80%. Nickel sulphate is available in various hydrate forms. The compositions reported here reflects the composition of the hydrated form of the material as placed on the market.

***This composition is neither manufactured nor imported but is only reported for technical reasons, as advised by ECHA. Analyses were performed on an anhydrous form generated from the marketed material using a drying process.